

INVENTOR:  
TITLE:  
ATTORNEY:  
EXPRESS NO.:  
SHEET 1 OF 6

DiFoggio  
A Method and Apparatus for Downhole Quantification...  
G. Michael Roebuck TELEPHONE NO.: 713-266-1130  
EV369817905US DOCKET NO.: 584-30872-US

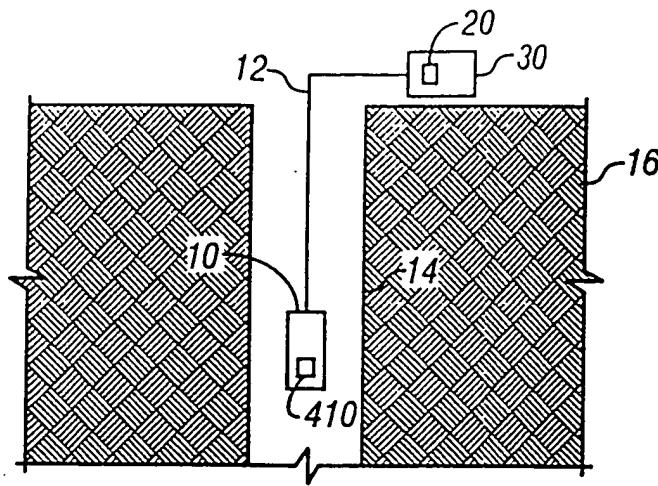


FIG. 1

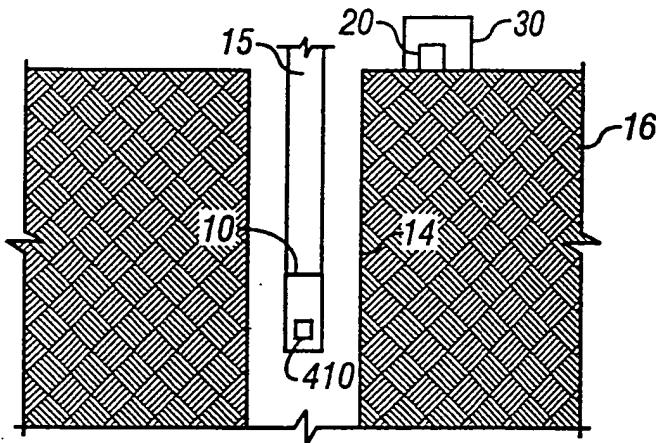


FIG. 2

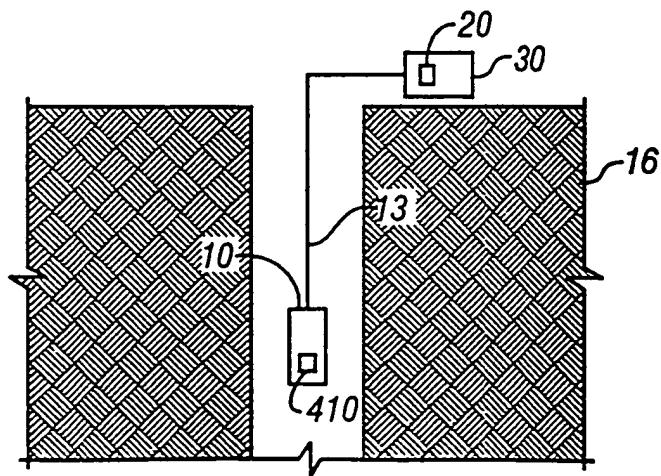


FIG. 3

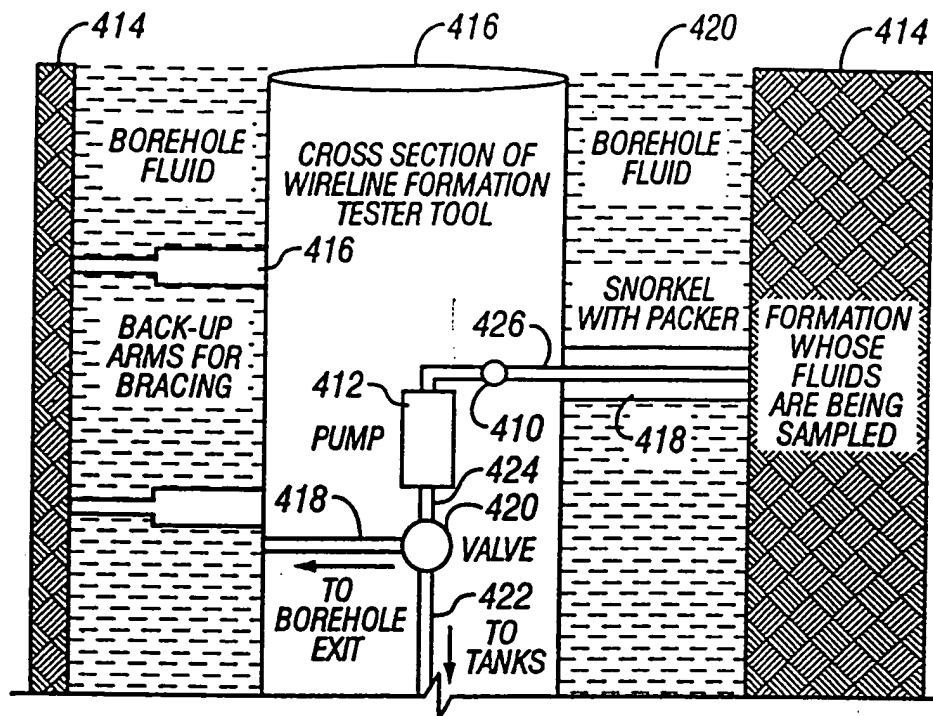


FIG. 4

INVENTOR:  
TITLE:  
ATTORNEY:  
EXPRESS NO.:  
SHEET 3 OF 6

DiFoggio  
A Method and Apparatus for Downhole Quantification...  
G. Michael Roebuck TELEPHONE NO.: 713-266-1130  
EV369817905US DOCKET NO.: 584-30872-US

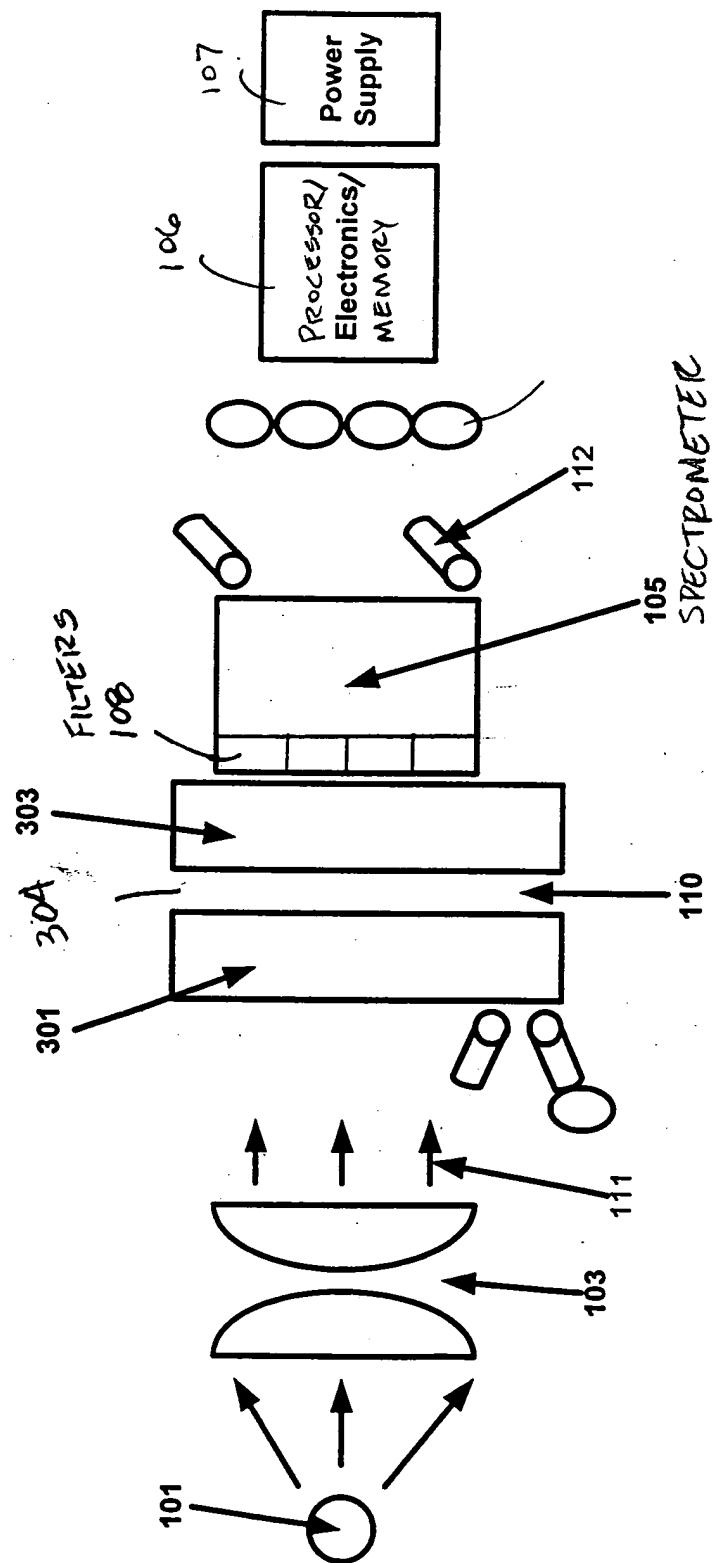


FIG. 5

**Equations Correlating Weight Fraction Methane  
in Mixtures of Crude Oil and Methane  
to Optical Absorbance and Temperature**

**FIG. 6**

Methane Weight Fraction =  $METHWTF = B0 + B1 * Var1 + B2 * Var2 + B3 * Var3 + B4 * Var4 \dots$

Regression Summary for Dependent Variable: **METHWTF**  
 $R = .98093203$   $R^2 = .96222765$  Adjusted  $R^2 = .9615158$   
 $F(4,211) = 1343.8$   $p < 0.0000$  Std.Error of estimate: .04992

**B**

$0.06514 = B0 = \text{Intercept}$

$11.1756 = B1$

$0.00087 = B2$

$-2.66167 = B3$

$2.63245 = B4$

**Var1 = SQ70\_82**

**Var2 = TEMP\_C**

**Var3 = SRSA1682**

**Var4 = SRSA1670**

SC70\_82 = SQUARE(Absorbance\_at\_1670\_nm - Absorbance\_at\_1682\_nm)

**SRSA1670 = SQRT(Absorbance\_at\_1670\_nm)**

**SRSA1682 = SQRT(Absorbance\_at\_1682\_nm)**

**TEMP\_C = Temperature in Degrees Centigrade**

**TEMP\_SQR = Square of Temperature in Degrees C**

Equation for Density of Methane [g/cc] as  
a Function of Pressure and Temperature  
from 100 - 30,000 psia and 75 - 200 C  
is fitted by

**B**

$0.03143 = B0 = \text{Intercept}$

$2.53111 = B1$

$-2.55766 = B2$

$11.9135 = B3$

$0.0019 = B4$

$-6.2E-06 = B5$

**P**

$2.480E-05 = \text{Intercept}$

$1.120E-09 = \text{for Pressure in psi}$

$P^2$

$1.808E-14 = \text{for Temperature in C}$

$P^3$

$1.455E-03 = \text{for Temperature in C}$

$T^2$

$-1.308E-07 = \text{for Temperature in C}$

$(P/T)$

$-4.922E-06 = \text{for Temperature in C}$

$(P/T)^2$

$5.934E-09 = \text{for Temperature in C}$

Equation for Optical Absorbance per mm of  
Methane as Function of Density and Wavelength  
at 11 nm FWHM, Center  $\lambda$  range of 1668-1684 nm,  
for 100-30,000 psia, and 75 - 200 C,  
is fitted by

**B**

$19.9061 = \text{Intercept}$

$0.7747 = \text{Methane Density}$

$3.3326 = \text{WaveNumber}/1000$

where, WaveNumber =  $10\ 000\ 000 / \lambda [\text{nm}]$

**GOR = Weight Fraction of Methane,  $f_M$ , and Stock Tank Density,  $\rho_0$ , of Oil**

$1 \text{ bbl} = 0.159 \text{ m}^3 \cdot 5.615 \text{ cu ft} = 42 \text{ U.S. gal}$

$1 \text{ Standard Cubic Foot (SCF)} \text{ of Methane Gas at } 14.7 \text{ psia \& } 60^\circ\text{F} \text{ is } 0.042358 \text{ lbs} = 19.21327 \text{ grams.}$

Density of Methane at 60 F and 14.7 psia is  $0.0006787 \text{ gr/cc} = 0.042358 \text{ lbm/ft}^3$ .

Letting  $V = \text{Volume}$ ,  $W = \text{Weight}$ ,  $\rho = \text{Density}$ , and using subscripts **M** for Methane and **O** for Oil,

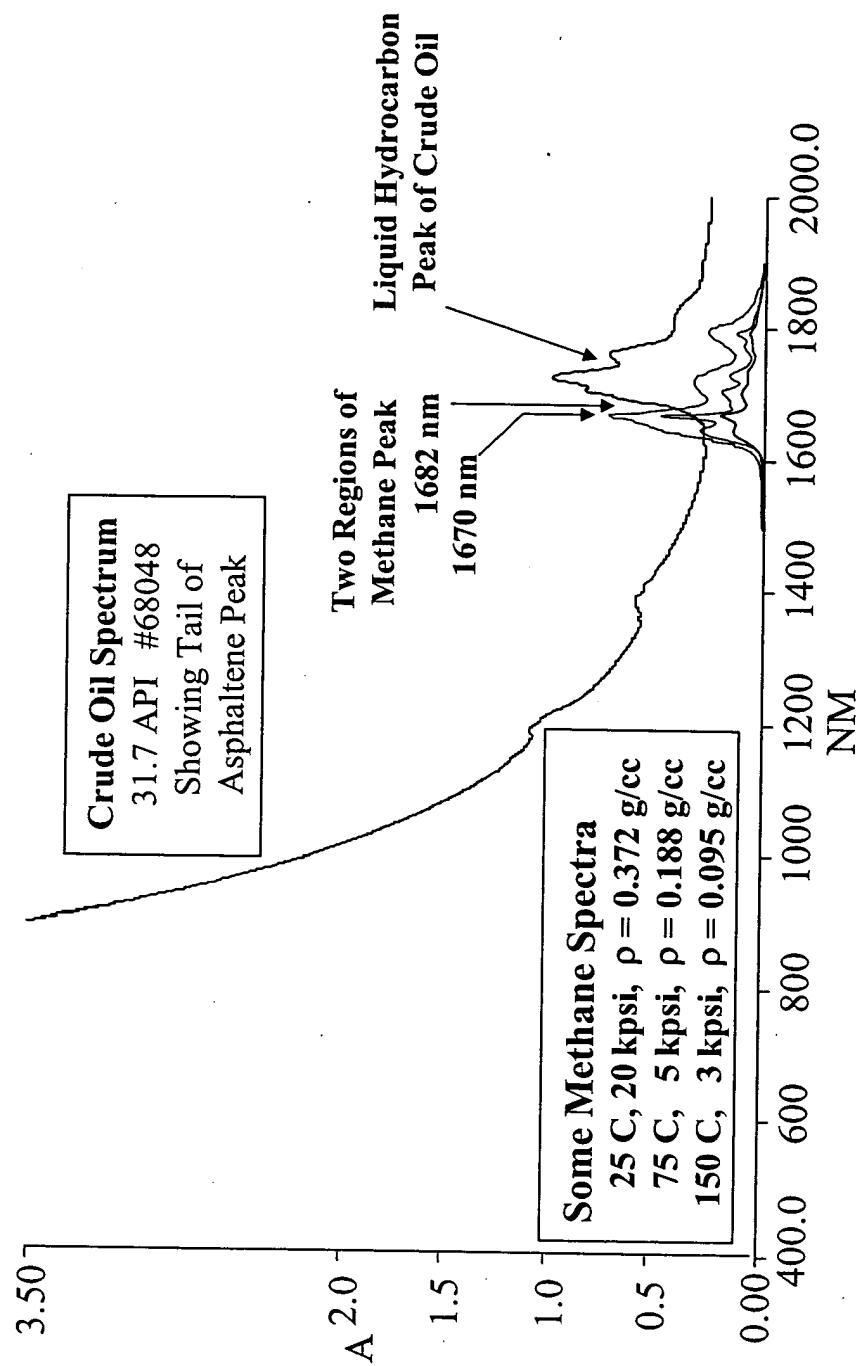
$GOR = V_{\text{Methane}} [\text{SCF}] / V_{\text{Oil}} [\text{bbls}] = \{W_M / (19.21 \text{ g/SCF})\} / \{(W_O / \rho_0) / (1 \text{ bbl} / 158.983 \text{ cc})\}$

Letting  $f_M = \text{Weight Fraction of Methane}$ ,

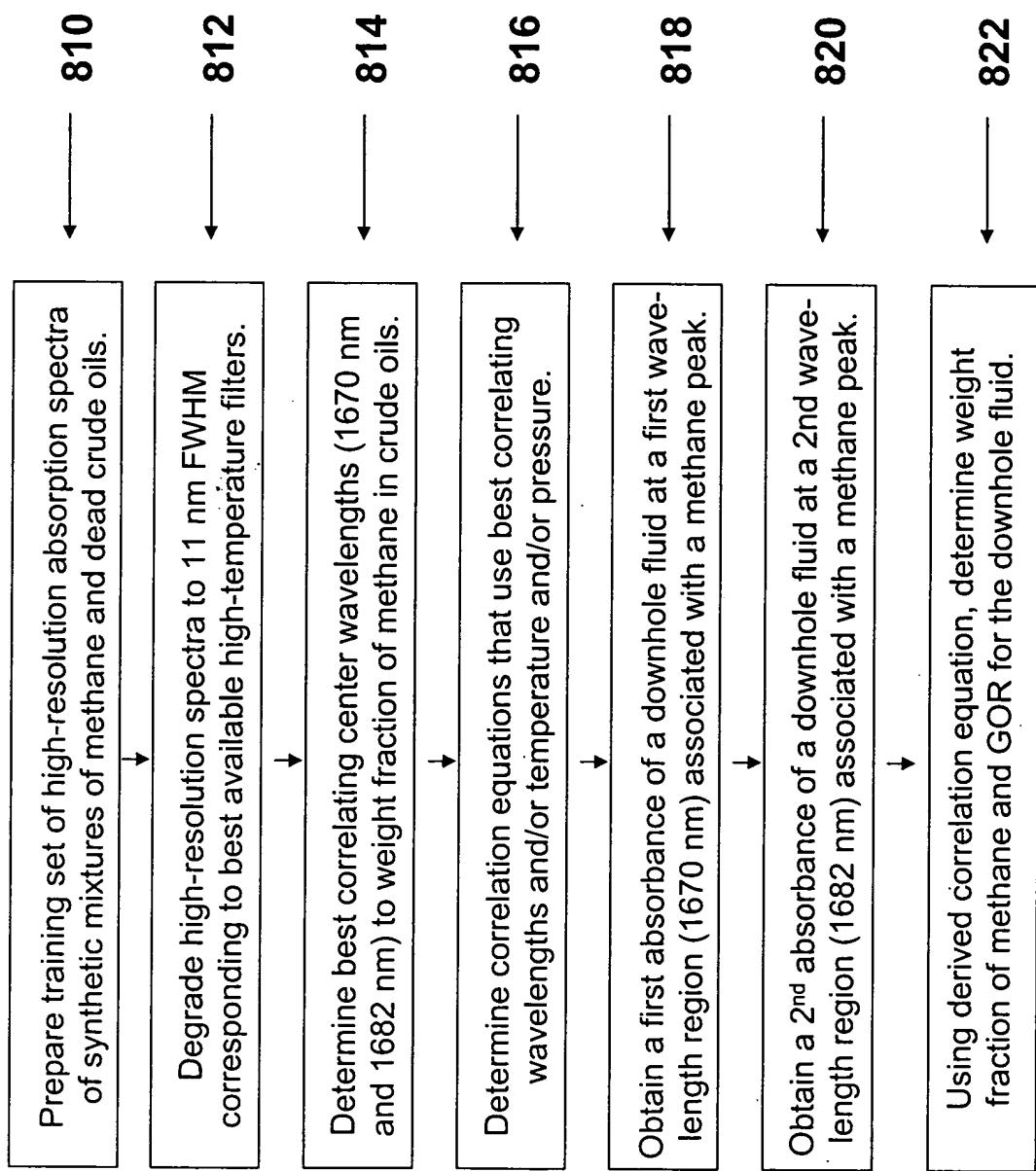
$GOR = 8274.62 \rho_0 / (1 / f_M - 1)$

$f_M = W_M / (W_M + W_O) = \rho_W V_M / (\rho_M V_M + \rho_O V_O) \text{ so } W_O = W_M / (1 / f_M - 1) \text{ which substitutes into above.}$

$f_M = 1 / (1 + 8274.62 * \rho_0 / GOR) \text{ where } W_G \text{ and } W_O \text{ are in grams, } \rho_0 \text{ is in g/cc, and } f_M = \text{Wt. Frac. of Methane}$



**Figure 7**



**Figure 8**